

## BRIEF NOTE

PROTECTIVE EFFECT OF ZINC CHLORIDE ON CARBON  
TETRACHLORIDE-INDUCED REDUCTION OF  
COMPLEMENT ACTIVITY IN RATS

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It has been demonstrated that total hemolytic complement activity in rats was markedly reduced following the administration of a sublethal dose of carbon tetrachloride (CCl<sub>4</sub>)<sup>1)</sup>. In other experiments, it has also been reported that CCl<sub>4</sub>-induced liver damage could be significantly prevented by pretreatment of animals with divalent metals, including zinc<sup>2,3)</sup> and cobalt<sup>4)</sup>. In this respect, a preliminary study was carried out to see whether zinc chloride (ZnCl<sub>2</sub>) would protect CCl<sub>4</sub>-induced reduction of total hemolytic complement activity in rats.

## MATERIALS AND METHODS

Male Wistar rats, 11 weeks old, were used in this study. The animals were fed Oriental MF pellets (zinc content 6.05mg/100g), and given tap water *ad libitum*. They were divided in 4 groups of 6 rats each. Rats were injected intraperitoneally with aqueous solution of ZnCl<sub>2</sub> (Kojundo Kagaku Kenkyusho) so that they received 250 μmol. Injection volume was 5 ml/kg and control animals received an equal volume of saline. Twenty-four hours following the pretreatment with ZnCl<sub>2</sub> or saline, rats were given intraperitoneally 1ml/kg of olive oil or 0.5ml/kg of CCl<sub>4</sub> in a 50% olive oil solution. Twenty-four hours later, the rats were sacrificed under light ether anesthesia and blood was drawn from abdominal aorta with plastic syringes. It was allowed to clot at room temperature for one hour, then left standing overnight at 4°C in a refrigerator, and serum was separated by centrifugation.

Total hemolytic complement activity in serum was assayed by a modified method of Mayer et al.<sup>5)</sup> as the 50 percent hemolytic unit (CH50). Total protein content of serum was determined by a modified biuret method (Wako). Composition of serum proteins was analyzed by the cellulose acetate membrane electrophoresis. Serum alkaline phosphatase (Al-P) was assayed by a modified method of Bessey-Lowry (Alkalinephospha B-Test, Wako). Serum glutamic oxalacetic transaminase (SGOT) was determined by a modified method of Reitmann and Frankel (S. TA-Test, Wako). Serum triglyceride level (TG) was assayed by the method of Moteki et al. (Triglyceride-Test, Wako). Concentration of zinc in serum was determined by a flame atomic absorption method (Perkin-Elmer 503)<sup>6)</sup>.

## RESULTS AND DISCUSSION

The results of the present investigation are summarized in Table 1,2 and Fig. 1.

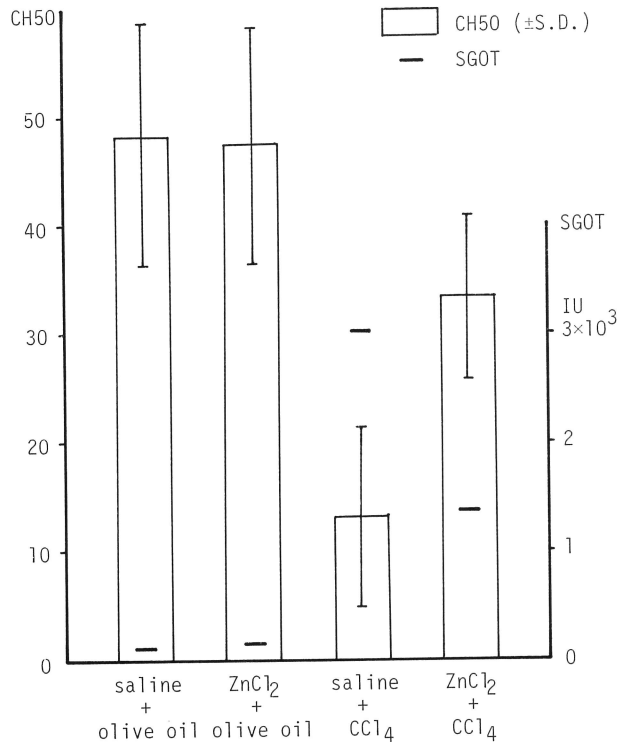


Fig. 1. Effect of ZnCl<sub>2</sub> on CCl<sub>4</sub>-induced alterations of complement and SGOT levels

Animals pretreated with ZnCl<sub>2</sub> had significantly higher CH50 values following CCl<sub>4</sub> than those of CCl<sub>4</sub>-treated animals, and the results demonstrated that pretreatment of rats with ZnCl<sub>2</sub> prior to CCl<sub>4</sub> resulted in significant protection against CCl<sub>4</sub>-induced reduction of total hemolytic complement activity (Table 1, Fig. 1). The results also suggested that the acute hepatic damage following CCl<sub>4</sub> intoxication, as measured by hepatic transaminase release (SGOT), could be partially prevented by pretreatment with ZnCl<sub>2</sub>, as reported by others (Table 2, Fig. 1). A close inverse correlation was observed between CH50 values and SGOT values, as an index of hepatic damage ( $r = -0.691$ ,  $p < 0.02$ ).

Protective effect of ZnCl<sub>2</sub> on CCl<sub>4</sub>-toxicity, as measured by serum TG, Al-P levels, composition of serum protein fractions, as well as relative weights of the livers and the kidneys, was not clearly observed (Table 2).

TABLE 1. Effect of ZnCl<sub>2</sub> on CCl<sub>4</sub>-induced reduction of complement activity

treatment	no. of animals	body weight g	CH 50	TP g/dl	A/G	serum Zn $\mu$ g/dl
saline+olive oil	6	324 $\pm$ 17 <sup>a)</sup>	47.6 $\pm$ 11.3	6.36 $\pm$ 0.18 <sup>**</sup>	0.94 $\pm$ 0.16	117 $\pm$ 22 <sup>**</sup>
ZnCl <sub>2</sub> +olive oil	6	308 $\pm$ 28	47.4 $\pm$ 10.9 <sup>**</sup>	5.60 $\pm$ 0.19 <sup>**</sup>	0.80 $\pm$ 0.07	302 $\pm$ 105 <sup>**</sup>
saline + CCl <sub>4</sub>	6	322 $\pm$ 9	13.0 $\pm$ 8.2 <sup>**</sup>	5.55 $\pm$ 0.27 <sup>**</sup>	0.80 $\pm$ 0.09	154 $\pm$ 32 <sup>**</sup>
ZnCl <sub>2</sub> + CCl <sub>4</sub>	6	326 $\pm$ 39	33.3 $\pm$ 7.6 <sup>**</sup>	4.91 $\pm$ 0.23 <sup>**</sup>	0.79 $\pm$ 0.11	349 $\pm$ 92 <sup>**</sup>

a) mean $\pm$ s. d. statistically significant \*p<0.05 \*\*p<0.01

TABLE 2. Effect of ZnCl<sub>2</sub> on CCl<sub>4</sub>-induced release of GOT into serum

treatment	no. of animals	GOT IU/l b)	TG mg/dl	Al-P BL c)	liver g/100g b. w.	kidneys g/100g b. w.
saline+olive oil	6	83 $\pm$ 22	79.4 $\pm$ 44.3	12.1 $\pm$ 5.0	3.62 $\pm$ 0.32	0.68 $\pm$ 0.04
ZnCl <sub>2</sub> +olive oil	6	124 $\pm$ 117	94.3 $\pm$ 32.3	8.6 $\pm$ 3.5	3.81 $\pm$ 0.17	0.67 $\pm$ 0.09
saline+CCl <sub>4</sub>	6	3023 $\pm$ 965 <sup>**</sup>	51.0 $\pm$ 31.0	11.4 $\pm$ 3.8	3.74 $\pm$ 0.48	0.68 $\pm$ 0.04
ZnCl <sub>2</sub> +CCl <sub>4</sub>	6	1382 $\pm$ 1307 <sup>**</sup>	47.4 $\pm$ 35.7	12.8 $\pm$ 6.6	3.84 $\pm$ 0.29	0.64 $\pm$ 0.04

b) International Unit c) Bessey-Lowry Unit

Carbon tetrachloride is well known to cause significant liver damage, and this damage is due to peroxidation of membrane lipids by reactive free radicals produced by the hepatic microsomal drug oxidizing system<sup>7)</sup>. It has been demonstrated that CCl<sub>4</sub>-induced liver injury could be prevented by pretreatment of animals with ZnCl<sub>2</sub>. The possible mechanism of the protective effect of ZnCl<sub>2</sub> has not been fully understood, however, divalent metal zinc was shown to be a stabilizer of biological membrane possibly because of its interference with lipid peroxidation<sup>8)</sup>. It has also been suggested that zinc-induced metallothionein could protect the CCl<sub>4</sub>-induced liver damage by sequestering reactive metabolites of CCl<sub>4</sub><sup>9)</sup>. However, the mechanism of CCl<sub>4</sub>-induced reduction of total hemolytic complement activity, as well as that of protective effect of zinc against the alteration of complement system have not been understood, and further studies should be needed to clarify these mechanisms.

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